Stereospecific exo-Incorporation of Deuterium into the Bicyclo[4.3.1] - decatetraenyl Anion

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Summary The title annulenyl anion (4) shows an overwhelming tendency for exo-face (cis to the methylene bridge) deuterium incorporation to give (5), contrary to a previous report describing the endo-deuteriation of (4).

We have recently demonstrated that the methoxycarbonyl bicyclo [4.3.1] decatetraenyl anion (1) shows a high exo-(cis to the methylene bridge)-stereoselectivity in its reaction with electrophiles giving the exo-incorporation products (2) (R=H, Me, and CH₂Ph). This selectivity of the anion (1) is the opposite to that obtained by Radlick and Rosen² for D₂O quenching of the anion (4) in Me₂SO. They indicated

that incorporation of deuterium into (4) occurred from the endo face because the ¹H n.m.r. signal at δ 2·71 of (3) disappeared in the deuteriated tricyclo[4.3.1.0]decatriene (5). It is, however, unlikely that the stereospecificity of the reaction with electrophiles is reversed according to whether an ester group is present or absent. Although Radlick and Rosen assigned the signals at δ 2·71 and 2·47† to H-9-endo and H-9-exo, respectively, in bicyclo[4.3.1.0]-decatriene (3), this assignment is rather ambiguous, since an anisotropic effect of the olefinic units is also conceivable in addition to the effect of the cyclopropane ring. For example, both the C-4 methylene protons of bicyclo[3.1.0]-hex-2-ene have nearly identical chemical shifts.

[†] The chemical shifts, δ 2.65 and 2.35, of the C-9 methylene protons reported in ref. 2 should be corrected as given here.

TABLE. ¹H N.m.r. data for (6) at 100 MHz, in CDCl₃

Assignment	$\delta/p.p.m.$	Multiplicity	$J/{ m Hz}$
H-7	3.40	dd	$J_{7,A}$ 5.0, $J_{7,8}$ 9.0
H-8	2.81	dddd	$J_{8,B} = 6.0$
H-9-endo	1.05	ddd	J10,9endo 2.0; J8,9endo 4.8
H-9-exo	2.01	dd	J ₈ , 9exo 9·0, J _{9exo} , 9endo 13·0
H-10-syn	1.20	d	I 10 sun 10 anti 3.5

To substantiate further the assignment of the ¹H n.m.r. spectra of the tricyclotriene (3) and to determine the mode of the deuteriation of (4), the triene (3) was allowed to react with isobenzofuran to afford a single adduct (6) [m.p. 73—75 °C; m/e 248 (3.7%), 130 (100), and 118 (75)]. Proof of the structure of (6) was based on its ¹H n.m.r. data, which are summarized in the Table. The values of $J_{7.4}$ and $J_{8,B}$ indicated a *cis*-arrangement of these pairs of protons, eliminating the two isomeric structures having a transarrangement of the corresponding protons. Of the two remaining structures, (6) was preferred to (7) on the basis of the following. (a) The chemical shift of H-10-syn in the adduct was comparable with shifts of the corresponding protons which are not shielded by the aromatic ring in tricyclo[4.3.1.0]deca-2,4-diene3 and its 8-endo-chloro4 derivative. (b) W coupling was observed between H-10-anti and the C-9 proton at δ 1.05 (J 2.0 Hz), but not H-7. (c) Of the two C-9 methylene protons, the resonance of that coupled with H-10-anti was ca. 1.0 p.p.m. to higher field than the non-coupled proton. This can be reasonably explained only by the structure (6) where H-9-endo is located in the shielding region of the aromatic ring.

Reaction of the monodeuteriated triene (5), obtained from the anion (4) by D₂O quenching, t with isobenzofuran gave the deuteriated adduct (8) [m.p. 75-76 °C; m/e 249 (1.9%), 131 (100), and 118 (52)]. The signal at δ 2.01 for (6) was absent from the spectrum of (8) and the signals at δ 1.05 (H-9-endo) and 2.81 for (6) collapsed to a lower multiplicity for (8), indicating unequivocally that (8) contains the deuterium in the C-9-exo position. In the two

double triplets centred at δ 2.71 and 2.47 for the C-9 methylene protons of (3), the former disappeared in the spectrum of deuteriated (5) and therefore should be assigned to H-9-exo in accordance with stereospecific exo-face incorporation of deuterium into the 1,5-methano[9]annulenyl anion (4).

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- ‡ Quenching with D₂O, AcOD, or MeOD of solutions of the lithium salt of the anion (4) in tetrahydrofuran, the sodium salt in dimethyl sulphoxide, and the potassium salt in 1,2-dimethoxyethane led to the same compound (5) in each case.
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